

Benzoin

Not to be confused with Benzoin resin.

Benzoin (/benzoɪn/ or /-ɔɪn/) is an organic compound with the formula PhCH(OH)C(O)Ph. It is a hydroxy ketone attached to two phenyl groups. It appears as off-white crystals, with a light camphor-like odor. Benzoin is synthesized from benzaldehyde in the benzoin condensation. It is chiral and it exists as a pair of enantiomers: (*R*)-benzoin and (*S*)-benzoin.

Benzoin is *not* a constituent of benzoin resin obtained from the **benzoin tree** (*Styrax*) or **tincture of benzoin**. The main component in these natural products is **benzoic acid**.

1 History

Benzoin was first reported in 1832 by **Justus von Liebig** and **Friedrich Woehler** during their research on oil of bitter almond, which is **benzaldehyde** with traces of hydrocyanic acid.^[2] The catalytic synthesis by the benzoin condensation was improved by **Nikolay Zinin** during his time with Liebig.^{[3][4]}

2 Uses

The main uses of benzoin are as a precursor to **benzil**, which is a photoinitiator.^[5] The conversion proceeds by **organic oxidation** using copper(II),^[6] nitric acid, or **oxone**. In one study, this reaction is carried out with atmospheric oxygen and basic alumina in dichloromethane.^[7]

3 Preparation

Benzoin is prepared from **benzaldehyde** via the **benzoin condensation**.^[8]

4 References

- [1] “Benzoin” (PDF). FischerSci. FischerSci.
- [2] Wöhler, Liebig; Liebig (1832). “Untersuchungen über das Radikal der Benzoësäure”. *Annalen der Pharmacie* **3** (3): 249–282. doi:10.1002/jlac.18320030302.

- [3] N. Zinin (1839). “Beiträge zur Kenntniß einiger Verbindungen aus der Benzoylreihe”. *Annalen der Pharmacie* **31** (3): 329–332. doi:10.1002/jlac.18390310312.
- [4] N. Zinin (1840). “Ueber einige Zersetzungprodukte des Bittermandelöls”. *Annalen der Pharmacie* **34** (2): 186–192. doi:10.1002/jlac.18400340205.
- [5] Hardo Siegel, Manfred Eggersdorfer “Ketones” in Ullmann’s Encyclopedia of Industrial Chemistry Wiley-VCH, 2002 by Wiley-VCH, Wienheim. doi:10.1002/14356007.a15_077
- [6] Clarke, H. T.; Dreger.E. E. (1941). “Benzil”. *Org. Synth.*; Coll. Vol. **1**, p. 87
- [7] Konstantinos Skobridis; Vassiliki Theodorou; Edwin Weber (2006). “A very simple and chemoselective air oxidation of benzoins to benzils using alumina”. *Arkivoc*. 06-1798JP: 102–106.
- [8] Clarke, H. T.; Dreger.E. E. (1941). “Benzil”. *Org. Synth.*; Coll. Vol. **1**, p. 87

5 External links

- Benzoin synthesis, *Organic Syntheses*, Coll. Vol. 1, p. 94 (1941); Vol. 1, p. 33 (1921)

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